SPECTRA AND STRUCTURE OF SOME FLOPPY MOLECULES COOLED IN SEEDED JET

Mihir Chowdhury Department of Physical Chemistry Indian Association for the Cultivation of Science Jadavpur, Kolkata 700 032, India

The pioneering works of Raman and Mizushima in the area of vibrational spectroscopy have been widely exploited by physical chemists for obtaining structural information about molecules. This lecture, dedicated to the two great scientists, reviews our work on low-frequency vibronic spectra of jet-cooled large organic molecules.

A chemical change involves hopping of the system from one well in the multidimensional potential energy surface (PES) to another. A simplified model for this process which lends itself to further detailed treatment is large-amplitude vibration along the soft mode of a floppy molecule. It is therefore pertinent that we collect more information about the ground and/or excited state PES corresponding to motions like aryl rotation, ring-puckering, H-bond stretching and intermolecular dissociation.

In the condensed phase the perturbation by the surrounding environment is large enough to cause deformation and blurring of conformer identities. It is necessary, therefore, to work with a set of collisionless, cold, uniformly moving molecules as provided in a jet. The vibronic features of the laser-induced fluorescence (LIF) and fluorescence excitation (FE) spectra of the cold molecular beam are often amenable to analysis in terms of a fundamental frequency and its higher harmonics, which in turn can be correlated with the nature of excited and ground PESs.

The molecules of interest to us are listed in the table. In particular, the power of twocolour IR-UV double resonance technique in determining structures of clusters is illustrated with five hydrates of tetrahydroisoquinoline.

Acknowledgement: The works have been carried out by my collaborators with DST support. I thank Professor O. Kajimoto and Prof. Y. Yoshihara for the invitation to deliver this talk. **References:**

- 1. T. Chakraborty, M.Chowdhury; Chem. Phys. Letts. 171, 1990, 25
- 2. N. Guchait, R. Dutta, P. Dutta, T. Chakraborty, Ind. J. Phys. 66B, 1992, 241
- 3. T. Chakraborty, M.Chowdhury; Chem. Phys. Letts. 177, 1991, 223
- 4. T. Chakraborty, D. N. Nath, M.Chowdhury; J. Chem. Phys. 96, 1992, 6456
- 5. N. Guchhait, T. Chakraborty, D.Majumdar, M.Chowdhury, J.Phy. Chem. 98,1994, 9227
- 6. A. Ckaraborty et al. J. Chem. Phys. 114, 2001, 865
- 7. S. Banerjee et.al. Chem. Phys. Letts. 366, 2002, 211
- 8. N. Guchhait et.al. J. Mol.Struct. 327, 1994, 161
- 9. A.Chakraborty et.al. J. Chem. Phys. 115, 2001, 1
- 10. N.Guchhait et. al. J. Chem. Phys. (Communicated)

Table I				
Molecules	Referen	Vibrational modes (cm ⁻¹)		Description of modes and
	-ce number	Ground	Excited	potential energy surface
	1	-	28	Butterfly (Puckering)
Dihydroanthracene				(symmetry-equivalent double well, low barrier)
Dinydroantinacene				
	2	-	53	Butterfly (Puckering)
				(symmetry-equivalent double well, low barrier)
Xanthene				
	3	-	90	Torsion
				(symmetry-equivalent double well, low barrier)
Dihydrophenanthrene				double well, low ballier)
F F	A	105	26	$\mathbf{D}_{\mathbf{r}} \mathcal{H}_{\mathbf{r}} \mathcal{H}_{\mathbf{r}} \left(\mathbf{D}_{\mathbf{r}} \right) = 1 \mathbf{h} \in \mathcal{H}_{\mathbf{r}}$
F F	4	135	26	Butterfly (Puckering) (soft mode with barrier less than
				zero-point energy)
F F				
Octafluoronaphthalene				
	5	105	95	Puckering Twist
		160	179	Puckering Bend
Tetralin			(()7	
	6	-	66, 87 128	Puckering (in phase and out-of-phase boat and chair
			128	form)
Hexahydropyrene				
	7	188	129	Ring Puckering twist
		81	60	[in phase and out-of-phase]
Octahydroanthracene				(Sym and antisym twist conformer)
	8		45	Puckering coupled to N-
	_		119	inversion
Indoline				
	9	95 144	92	
Tetrahydroisoquinoline		144	98	Out-of-plane N-puckering (axial and equatorial form)
	10	3717		Free OH
.H2O	10	3717 3377	-	H-bonded OH
		3352	-	H-Bonded NH
THIQ $(H_2O)_1$ cluster				(Five clusters with five
				distinct IR-signatures)