# 3P052

## Structure and charge transport properties of tin-iodide perovskites

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## Introduction

Tin and iodide can be employed to construct a series of perovskites; the composition is denoted by  $A_2(Me-NH_3)_{n-1}Sn_nI_{3n+1}A_2Sn^{II}I_4$ , where A=organic ammonium cation, and the number of the layer (*n*) can be adjusted. The compounds with n = 1 are primarily semiconductors with wide energy gaps ( $E_g > 1 \text{ eV}$ ), but the electrical conductivity was found to be tunable by doping treatment (Sn(II) is partly replaced with Sn(IV)). Therefore, relatively high conductivity of as-grown  $A_2SnI_4$  was interpreted by spontaneous hole-doping [1]. It should be noted that even Me-NH<sub>3</sub>SnI<sub>3</sub> ( $n=\infty$ ) is also a semiconductor as shown by recent theoretical calculations, though the pressed powder sample showed metallic conductivity [2]. Therefore, the high conductivity of cubic perovskite Me-NH<sub>3</sub>SnI<sub>3</sub> is also considered to be due to spontaneous hole doping. In the present study, doping effects on the transport properties in the double layer Sn-I perovskite, (PhEtNH<sub>3</sub>)<sub>2</sub>(Me-NH<sub>3</sub>)Sn<sub>2</sub>I<sub>7</sub> (n=2) and cubic perovskite Me-NH<sub>3</sub>SnI<sub>3</sub> ( $n=\infty$ ) have been investigated.

#### Experiment

The double layer perovskite of  $(PhEtNH_3)_2(Me-NH_3)Sn_2I_7$  was prepared by adding anhydrous ethanol to stoichiometric amounts of purified SnI<sub>2</sub>, PhEtNH<sub>3</sub>I, and Me-NH<sub>3</sub>I under nitrogen atmosphere, followed by slow cooling of the solution (from 65 °C to 5 °C at a rate of 2 °C h<sup>-1</sup>). Black aggregated platelet crystals with size as large as  $1.5 \times 1.5 \times 1 \text{ mm}^3$  were obtained. The same method was employed to prepare Me-NH<sub>3</sub>SnI<sub>3</sub> (black and block crystals with size  $1.2 \times 1.2 \times 1 \text{ mm}^3$ ) by just using stoichiometric SnI<sub>2</sub> and Me-NH<sub>3</sub>I.

### **Results and analysis**

According to the single crystal X-ray diffraction at 153 K, (PhEtNH<sub>3</sub>)<sub>2</sub>(Me-NH<sub>3</sub>)Sn<sub>2</sub>I<sub>7</sub> has the space group of *P*-1 with cell parameters: a = 8.683, b = 8.659, c = 22.409 Å,  $\alpha = 93.74$ ,  $\beta$ = 97.42,  $\gamma = 90.24^{\circ}$ . Although the diffraction data of the crystal were not good enough, its double-layer feature was confirmed as shown in Fig. 1. The PhEtNH<sub>3</sub><sup>+</sup> ions are located between the tin-iodide perovskite double layers while Me-NH<sub>3</sub><sup>+</sup> ions are embedded into the intra-layer. The band calculation based from the inorganic framework indicates its semiconducting electronic structure with an approximate  $E_g$  of 1.0 eV. The primary electrical conductivity experiment (four-probe method) indicates that (PhEtNH<sub>3</sub>)<sub>2</sub>(Me-NH<sub>3</sub>)Sn<sub>2</sub>I<sub>7</sub> has the resistivity of 30-75  $\Omega$ · cm at 300K, comparable with that of the single layer perovskite (PhEtNH<sub>3</sub>)<sub>2</sub>SnI<sub>4</sub>[1].

In the following work, we will further carry out the electrical conductivity measurements of the single crystals of these two perovskites and their artificial Sn(IV)-doping, and investigate the relationship between the conductivity and the doping level.



Fig.1 double layer configuration (left) of the  $(PhEtNH_3)_2(Me-NH_3)Sn_2I_7$  (the PhEtNH<sub>3</sub><sup>+</sup> cations are not well resolved) and its band structure (right).

## References

[1] Y. Takahashi, R. Obara, K. Nakagawa, M. Nakano, J. Tokita, and T. Inabe: *Chem. Mater.*, 2007, 19, 6312-6316.

[2] D. B. Mitzi, C. A. Field, Z. Schlesinger, and R. B. Laibowitz, J. Solid State Chem., 1995, 114, 159.