## Molecular and Electronic Structures of Di-erbium and

## Di-erbium-carbide Metallofullerenes Er<sub>2</sub>(C<sub>2</sub>)@C<sub>82</sub>: Density Functional

## **Theory Calculations**

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Density functional theory geometry calculations have been performed to elucidate the molecular and electronic structures of di-erbium and di-erbium-carbide endohedral metallofullerenes  $\text{Er}_2@C_{82}$  and  $\text{Er}_2C_2@C_{82}$ .  $C_s(6)$  (I),  $C_{2\nu}(9)$  (II), and  $C_{3\nu}(8)$  (III) IPR isomers of C<sub>82</sub> were considered. We have employed B3LYP using Ahlrich's SVP and TZVPP basis sets for carbon, and the Stuttgart-Dresden SRSC97-ECP effective core potential and basis sets both with explicit consideration of f electrons and without, assuming in the latter case a  $6s^25d^1$ valence configuration to emulate the effects of strong electron correlation. The optimized structures are shown in Figure 1. We found that in each compound the erbium atom assume a trivalent state, where two electrons from each Er are transferred to the C<sub>82</sub> cage (LUMO and LUMO+1 become doubly occupied, corresponding to the  $C_{82}^{4-}$  electronic state), and one is either transferred to C<sub>2</sub> (erbium carbide compounds) or engages in covalent Er-Er and Er-C bonding (erbium-only compounds). Such chemical bonding situation is comparable to that encountered in Er<sub>3</sub>N@fullerene compounds, where the cages are found to be 6-fold negatively charged, and three covalent Er-N bonds exist. In particular, the C2 unit engages in strong covalent binding to the Er atoms, which gives rise to the notion of only weak charge uptake as detected by MEM-Rietveld X-ray measurements. Since the Er 4f shell is occupied by only about eleven electrons according to Mulliken population analysis, Er<sup>3+</sup>-like emission from  ${}^{4}I_{13/2}$  to  ${}^{4}I_{15/2}$  is consistent with the notion of only 2 electrons per Er fully transferred to the cage. The higher photoluminescent activity of isomer III compared to other isomers I and II stems from a sizable (LUMO+2)-(LUMO+1) gap in the  $C_{3\nu}(8)$  cage. Our prediction for photoluminescence intensity for all three isomers is consistent with the experimental results by Shinohara and coworkers [1].



Figure 1: Optimized structures for Er<sub>2</sub>@C<sub>82</sub> (I II III) and Er<sub>2</sub>C<sub>2</sub>@C<sub>82</sub> (I II III)

## Reference

 Ito Y., Okazaki T., Okubo S., Akachi M., Ohno Y., Mizutani T., Nakamura T., Kitaura R., Sugai T., and Shinohara H. Enhanced 1520 Photoluminescence from Er3+ Ions in Di-Erbium-carbide Metallofullerenes (Er<sub>2</sub>C<sub>2</sub>)@C<sub>82</sub> (Isomers I, II, and III), ACS Nano, 1(5), 456-462 (2007).