

# 1P041 Study of the Wavepacket Motion in Photoisomerization of Cyanine Dye by Ultrafast Pump-Dump-Probe Technique

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## 【Introduction】

The dynamics of chemical reactions is usually described by the rate of population transfer between two states corresponding to the local minima on the potential energy surface (PES). In ultrafast reactions, however, we can see the dynamics that is regarded as a fast and continuous change within a relevant excited-state PES. Figure 1 depicts continuous motions of a wavepacket, a superposition of states having different energies, on a monotonously-sloping barrierless  $S_1$  PES along the simplified reaction coordinate. Although this wavepacket provides an intuitive picture of the dynamics of ultrafast reactions, in reality, we only have very limited knowledge on how the wavepacket evolves on the PES. Therefore, it is highly desirable to experimentally examine and “visualize” the motion of the wavepacket, thereby unveiling the nature of the ultrafast reaction. In this presentation, we show our pump-dump-probe spectroscopic study of the excited-state wavepacket motion in ultrafast photoisomerization of a cyanine dye molecule.

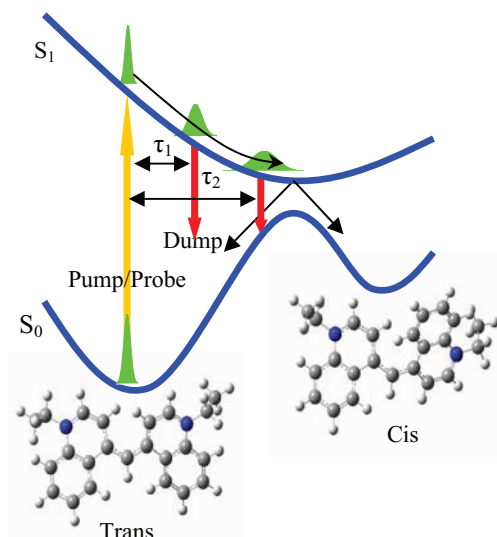


Fig. 1 Schematic illustration of the  $S_0$  and  $S_1$  PES as well as the wavepacket motion monitored by pump-dump-probe experiments.

## 【Results and discussion】

The cyanine dye (1144-C), which we selected as a sample, is known as a representative molecule showing ultrafast photoisomerization [1]. As shown in Figure 1, it has been suggested that the wavepacket generated by the pump pulse (580 nm) moves rapidly on the  $S_1$  PES from the Franck-Condon state toward a sink region, where the wavepacket goes back to the ground state with a time constant of  $\sim 9$  ps. This recovery of the ground-state bleaching is readily seen by ordinary pump-probe measurements (Figure 2 black curve).

To examine the wavepacket motion on the  $S_1$  PES, we introduced a dump pulse ( $\lambda_{\text{dump}}=690$  nm) that spectrally overlaps the stimulated emission band. As shown in Figure 2, the bleaching signal decreased instantaneously at each dump time, indicating that a substantial fraction of the  $S_1$  molecules was driven back to the ground state through the stimulated emission transition by the dump pulse. We evaluated the dump efficiency which was defined as,

$$\text{Dump efficiency} = 1 - \frac{\Delta\text{abs}(\text{dump on})}{\Delta\text{abs}(\text{dump off})}$$
, for every dump time, and plotted it in Fig. 3a. Interestingly, the dump efficiency showed a rise time with  $\sim 100$  fs time constant. This result clearly indicates that the wavepacket needs this finite time to reach the potential region, where the  $S_1$ - $S_0$  energy difference matches the photon energy of dump pulse at 690 nm. Furthermore, when the photon energy of the dump pulse was lowered ( $\lambda_{\text{dump}}=950$  nm), the corresponding time

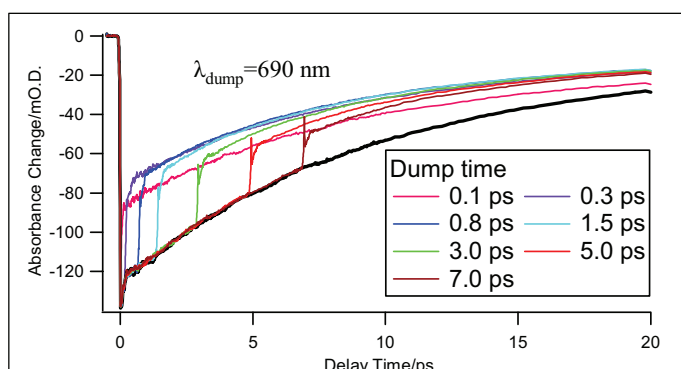


Fig. 2 Time-resolved pump-dump-probe signals obtained with and without the dump pulse for various pump-dump delays (1 mM solution in ethylene glycol, pump/probe: 580 nm, dump: 690 nm).

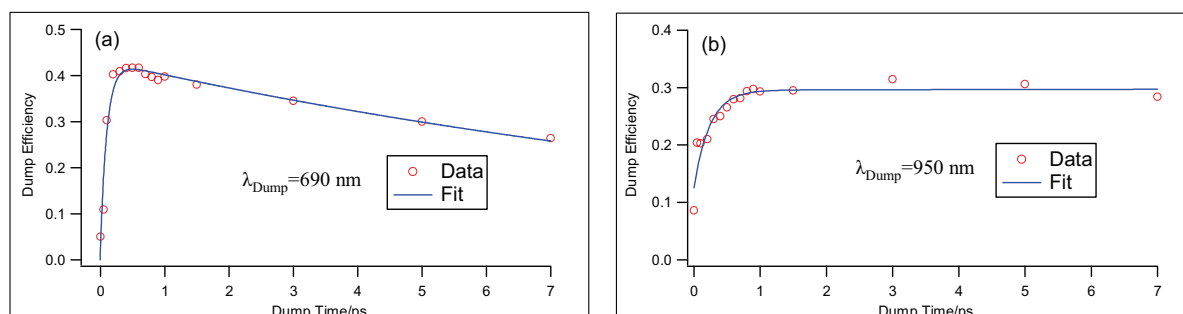


Fig. 3 Dump efficiency as a function of the dump time for the dump wavelength of (a) 690 nm and (b) 950 nm.

became substantially longer ( $\sim 240$  fs). This corresponds to the change of the potential region where the dump process occurs: the 950 nm pulse dumps the  $S_1$  state in the vicinity of the sink where the  $S_1$  and  $S_0$  energies come closer to each other (Figure 3b). The present observation provides a quantitative representation of the wavepacket motion as well as a relative shape of the  $S_0$  and  $S_1$  PES along the reaction coordinate.

## 【References】

[1]. E. Akesson, H. Bergstrom, V. Sundstrom, T. Gillbro, Chem. Phys. Lett. **126**, 385 (1986).